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The first syntheses of 6,7-dihydroxylated calystegines and homocalystegines

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Abstract—The synthesis of novel calystegine analogues containing hydroxyl substituents on both C6 and C7 positions has been achieved. The synthetic strategy employed is based on stereoselective dihydroxylation reactions and has also been utilised for the preparation of homocalystegines.

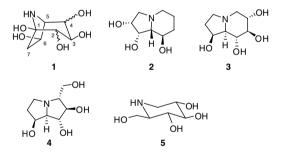
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Calystegines 1 are polyhydroxylated nortropane alkaloids that were first isolated by Tepfer et al. from the roots and root extrudates of *Calystegia sepium*.¹ They are rigid azasugar derivatives and have bicyclic structures that are similar to swainsonine 2, castanospermine 3 and australine 4. Azasugars are often strong inhibitors of glycosidase enzymes and have consequently been investigated as potential therapies in the treatment of diabetes, HIV and cancer.² In common with these biological properties, calystegines have also been shown to be selective glycosidase inhibitors. For example, calystegine B₂, which has a hydroxylation pattern resembling that of D-glucose, shows 100-fold higher inhibition of β -glycosidases from almonds and *Caldocellum saccharolyticum* than deoxynojirimycin 5.³

A number of naturally occurring calystegines have been isolated (and synthesised) to date.⁴ These compounds have been found to possess between three to five hydroxyl groups around the bicyclic core. However, in spite of this variable hydroxylation pattern, none of them contain hydroxyl groups on both C6 and C7. In addition, only a small number of non-natural calystegines have been synthesised and few have been tested with respect to their biological activity.⁵ This prompted us to explore an easily adaptable synthetic route to non-natural calystegines in order to further elucidate the structure–activity profile of this class of compounds as glycosidase inhibitors. This letter presents our initial results con-

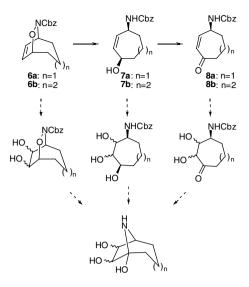
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cerning the synthesis of novel 6,7-dihydroxylated calystegines and the corresponding homocalystegine analogues.



Our synthetic strategy towards non-natural calystegines and homocalystegines is outlined in Scheme 1. The known bicyclic starting materials $6a^6$ and $6b^7$ can be readily accessed via cycloaddition of a nitroso-compound (generated in situ from the reaction of benzyl *N*-hydroxycarbamate with tetra-*n*-butyl-ammonium periodate)⁸ with cycloheptadiene or cyclooctadiene, respectively. We envisaged that the known conversion^{6,7} of **6a–b** to the corresponding alcohols **7a–b** and ketones 8a-b would allow us to investigate the stereoselective dihydroxylation of each of these three species with a view to access both possible syn-diols at C6 and C7. Although the syntheses described below are racemic, it should be noted that an enantioselective version of the cycloaddition reaction to produce 6a-b is possible thus allowing the methods reported here to be used for the production of homochiral material.9

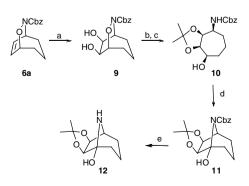
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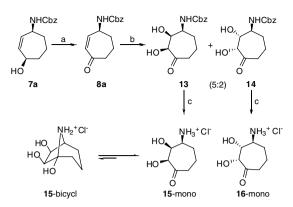
Scheme 1.

We started our investigations with the seven-membered calystegine series. Dihydroxylation of 6a under Upjohn conditions gave the β -diol 9 exclusively in good yield and with no evidence of attack from the α -face (Scheme 2).¹⁰ The stereostructure of **9** was determined by NOESY spectroscopy which indicated strong cross peaks between the C6 and C7 hydrogen atoms and the C3 α -hydrogen.^{11,12} Whilst the stereoselectivity for the dihydroxylation of **6a** is in agreement with the *exo*-attack observed for bicyclo[2.2.1]heptenes,¹³ it should be noted that an analogous bicyclo[2.2.2]oxazine substrate reportedly undergoes endo-selective dihydroxylation.¹⁴ Protection of the diol 9 followed by N–O bond cleavage using samarium diiode¹⁵ gave the alcohol 10. Dess-Martin oxidation of 10 then generated the corresponding ketone which existed predominantly as the hemiaminal 11 (>90% by NMR). Removal of the Cbz group to give 12 proved straightforward, but unfortunately all efforts to remove the isopropylidene protecting group from 12 using acidic catalysis met with failure due to significant decomposition of the starting material and/or product under the reaction conditions.

In view of the failure to effect the deprotection of **12** we turned our attention to the dihydroxylation reactions of



Scheme 2. Reagents and conditions: (a) OsO_4 , NMO, acetone/H₂O, rt, 48 h, (76%); (b) dimethoxypropane, *p*-TSA, acetone, rt, 72 h, (67%); (c) SmI₂, THF, rt, 1 h, (88%); (d) Dess–Martin periodinane, CH₂Cl₂, rt, 18 h, (48%) and (e) Pd(OH)₂, H₂, EtOAc, 18 h, (99%).

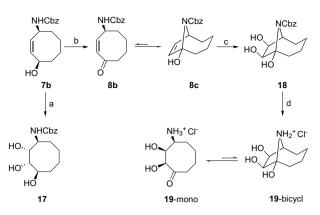


Scheme 3. Reagents and conditions: (a) Jones reagent, acetone, $0 \,^{\circ}$ C, 5 min (62%); (b) OsO₄, NMO, acetone/H₂O, rt, 24 h (13–51%; 14–20%) and (c) Pd(OAc)₂, H₂, EtOH:HOAc 18:1, rt; then H₂O/HCl (15-100%; 16–68%).

7a and 8a (Scheme 3). The alcohol 7a was readily accessed from 6a by N–O bond cleavage using $Mo(CO)_{6}$.¹⁶ However, *syn*-dihydroxylation of 7a under Upjohn conditions was non-selective and gave an inseparable (1:1) mixture of diastereoisomeric triols. A similar lack of facial selectivity has previously been observed in the epoxidation of 7a.¹⁷

Jones oxidation of **7a** produced the ketone **8a**¹⁸ in good yield. Pleasingly, dihydroxylation of **8a** proved to be more facially selective than the reaction of **7a** and afforded a mixture of **13** and **14** (ratio ca. 5:2) which could be separated by careful column chromatography and recrystallisation.¹⁹ Subsequent deprotection of **13** cleanly generated the novel calystegine analogue **15**. Interestingly, NMR analysis indicated that **15** exists predominantly in its monocyclic form rather than as the hemiaminal (**15**-mono:**15**-bicycl = 10:1).²⁰ This is in contrast to the majority of naturally occurring calystegines that have been isolated.⁴ In an analogous manner, deprotection of **14** gave the novel calystegine analogue **16** which again exists mainly as the monocyclic isomer (**16**-mono:**16**-bicycl = 2:1).²¹

Having successfully accessed the novel calystegine analogues 15 and 16 we next examined the dihydroxylation reactions of 6b-8b with a view to produce the corresponding homocalystegines. However, in contrast to the reaction of **6a**, the dihydroxylation of the bicyclic oxazine 6b led only to a mixture of unidentifiable products. The reaction of alcohol 7b (produced from 6b)⁷ was more successful, affording 17 as a single stereoisomer (Scheme 4).²² The increased facial selectivity in the dihydroxylation reaction of 7b compared to 6b has been noted before in the corresponding epoxidation of these substrates.^{17,23} Disappointingly, all attempts to protect the *syn*-diol in **17** led only to selective protection of the *anti*-1,2-diol functionality,²⁴ thus limiting the effectiveness of this route. Oxidation of the alcohol 7b gave the corresponding ketone as a separable mixture of keto-8b and hemiaminal-8c tautomers (8b:8c = 1:2).^{7,25} Gratifyingly, dihydroxylation of the bicyclic tautomer 8c proceeded in a stereoselective manner to afford the triol 18 in good yield.²⁶ Subsequent deprotec-



Scheme 4. Reagents and conditions: (a) OsO_4 , NMO, acetone/H₂O, rt, 72 h (72%); (b) Jones reagent, acetone, 0 °C, 5 min (77%–**8b:8c**; 1:2); (c) OsO_4 , NMO, acetone/H₂O, rt, 72 h (68%); (d) Pd(OAc)₂, H₂, EtOH:HOAc 18:1, rt; then H₂O/HCl (68%).

tion gave the novel homocalystegine 19 which again exists mainly as the monocyclic isomer (19-mono:19-bicycl = 10:3).²⁷

In summary, we have reported the first synthesis of nonnatural calystegines 15 and 16 and the first homocalystegine 19 in racemic form. Surprisingly, we have found that these natural product analogues exist mainly as the monocyclic keto-tautomer which is in contrast to the bicyclic hemiaminal-form reported for the majority of the calystegines isolated to date. Biological testing of 15, 16 and 19 produced in the course of this work is currently underway and these results together with the synthesis of further calystegine analogues will be reported in due course.

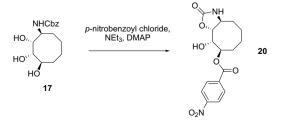
Acknowledgement

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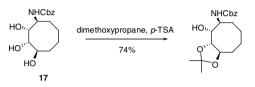
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- For safety reasons we replaced the Me₄NIO₄ used in the original preparations of **6a** and **6b** with *n*-Bu₄NIO₄—see Malpass, J. R.; Hemmings, D. A.; Wallis, A. L.; Fletcher, S. R.; Patel, S. J. Chem. Soc., Perkin Trans. 1 **2001**, 1044–1050.

- 9. Faitg, T.; Soulie, J.; Lallemand, J-Y.; Ricard, L. Tetrahedron: Asymmetry 1999, 10, 2165–2174.
- 10. The α and β -faces are defined as used in tropane systems.
- 11. All new compounds gave satisfactory spectroscopic and microanalytical and/or HRMS analysis.
- 12. For **9**, $\delta_{\rm H}$ (300 MHz, CDCl₃): 1.30–1.47 (1H, m, α -H3), 1.67–1.74 (1H, m, β -H3), 1.85–1.97 (3H, m, 2×H4 & 1×H2), 2.01–2.10 (1H, m, 1×H2), 3.91 (1H, d, J = 8.3 Hz, H7), 4.03 (1H, dd, J = 8.3, 1.8 Hz, H6), 4.48 (1H, dd, J = 3.8, 1.8 Hz, H1), 4.58 (1H, br s, H5), 5.21 (2H, s, CH₂–Ph), 7.29–7.40 (5H, m, Ph). $\delta_{\rm C}$ (75 MHz, CDCl₃) 19.1 (CH₂), 28.9 (CH₂), 29.8 (CH₂), 57.5 (CH), 65.1 (CH), 65.8 (CH), 67.7 (CH₂), 82.0 (CH), 128.1 (CH), 128.2 (CH), 128.5 (CH), 136.1 (C), 155.5 (C). Accurate mass (FAB): found: 294.1342 (M+H⁺); calcd. for C₁₅H₂₀NO₅: 294.1342.
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- 18. Ketone **8a** exists solely as the monocyclic tautomer shown—see Ref. 6.
- 19. Compound 13 exists solely as the monocyclic ketotautomer shown. The stereostructure of 13 was confirmed by protection of the diol as the isopropylidene acetal to give 11 and comparison to the material synthesised by the route shown in Scheme 2.
- 20. For **15**-mono (assignments based on calystegine numbering); $\delta_{\rm H}$ (300 MHz, D₂O): 1.52–1.75 (2H, m, 1×H3 & 1×H4), 1.80–1.95 (2H, m, 1×H3 & 1×H4), 2.30 (1H, dddd, J = 2.1, 6.6, 10.0, 12.8 Hz, 1×H2), 2.57 (1H, dt, J = 12.8, 3.5 Hz, 1×H2), 3.50–3.59 (1H, m, H5), 4.04 (br s, 1H, H6), 4.70 (1H, H7 overlap with D₂O); $\delta_{\rm C}$ (75 MHz, D₂O) 17.7 (CH₂), 26.4 (CH₂), 40.2 (CH₂), 53.9 (CH), 72.8 (CH), 77.1 (CH), 212.5 (C); accurate mass (FAB): found: 160.0973 (M–Cl⁻); calcd. for C₇H₁₄NO₃: 160.0974. Selected signals for **15**-bicycl; $\delta_{\rm H}$ (300 MHz, D₂O): 3.99 (1H, d, J = 6.9 Hz, H6 or H7), 4.19 (1H, d, J = 6.9 Hz, H6 or H7), all other ¹H signals overlap with **15**-mono; $\delta_{\rm C}$ (75 MHz, D₂O) 16.2 (CH₂), 24.1 (CH₂), 32.4 (CH₂), 61.9 (C), 69.8 (CH), 70.9 (CH), 92.8 (C).
- 21. For **16**-mono (assignments based on calystegine numbering); $\delta_{\rm H}$ (300 MHz, D₂O): 1.50–2.06 (4H, m, 2×H3 & 2×H4), 2.46 (1H, dddd, J = 16.5, 15.2, 8.1, 7.2 Hz, 1×H2), 2.65 (1H, dt, J = 16.5, 5.6 Hz, 1×H2), 3.45 (1H, dt, J = 3.2, 8.7 Hz, H5), 3.88 (1H, dd, J = 8.7, 2.4 Hz, H6), 4.47 (1H, d, J = 2.4 Hz, H7); $\delta_{\rm C}$ (75 MHz, D₂O) 18.0 (CH₂), 27.6 (CH₂), 39.3 (CH₂), 54.7 (CH), 71.9 (CH), 78.5 (CH), 214.9 (C); accurate mass (FAB): found: 160.0975 (M-Cl⁻); calcd. for C₇H₁₄NO₃: 160.0974.
- 22. The stereostructure of **17** was confirmed by X-ray crystallographic analysis of the derivative **20**.



Crystal data for compound **20**: $C_{16}H_{18}N_2O_7$, M = 350.32, triclinic, space group *P*-1, a = 7.082(2), b = 7.129(2), c = 17.347(5) Å, V = 781.4(4) Å³, T = 150(2) K, Z = 2, Dc = 1.489 g cm⁻³, μ (Mo-K α) = 0.118 mm⁻¹. Final, $R_1 = 0.0518$ (for 5693 reflections with $I > 2\sigma(I)$) and wR2 = 0.0642 for all data. Crystallographic data has been deposited with the Cambridge Crystallographic Data Centre as supplementary publication number CCDC 617411. Copies of the data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK (fax: +44 1223 336033 or e-mail: deposit@ccdc.cam.ac.uk).

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- 24. For example:



- 25. Once separated **8c** appears to be stable and does not reequilibrate with **8b** to any appreciable extent after standing for 24 h at room temperature in CDCl₃. Storage of ketone **8b** under these conditions, however, re-establishes the 1:2 ratio of **8b:8c**.
- 26. Triol 18 exists mainly as the hemiaminal tautomer shown.
- 27. For **19**-mono (assignments based on calystegine numbering); $\delta_{\rm H}$ (300 MHz, D₂O): 1.31–2.04 (6H, m, 2×H3, 2×H4 and 2×H5), 2.26–2.34 (1H, m, 1×H2), 2.73 (1H, dt, J = 3.5, 12.8 Hz, 1×H2), 3.44–3.51 (1H, m, H6), 4.55 (1H, d, J = 2.5 Hz, H8), 4.69 (1H, H7, overlap with D₂O); $\delta_{\rm C}$ (75 MHz, D₂O) 22.6 (CH₂), 26.6 (CH₂), 35.8 (CH₂), 38.2 (CH₂), 54.9 (CH), 72.5 (CH), 78.7 (CH), 215.7 (C); accurate mass (FAB): found: 174.1130 (M–Cl⁻); calcd. for C₈H₁₆H₁₆NO₃: 174.1130. Selected signals for **19**bicycl; $\delta_{\rm H}$ (300 MHz, D₂O): 3.65–3.67 (1H, m, H6), 4.12 (1H, d, J = 5.3 Hz, H8), 4.24 (1H, dd, J = 2.5, 5.3 Hz, H7), all other ¹H signals overlap with **19**-mono; $\delta_{\rm C}$ (75 MHz, D₂O) 20.4 (CH₂), 21.7 (CH₂), 26.8 (CH₂), 27.5 (CH₂), 60.4 (C), 66.6 (CH), 73.9 (CH), 95.5 (C).